## **STATES OF MATTER**

A. Matter can be categorised into the three states which can change their states when they experience a change in temperature and/or pressure.

	solid - liquid – vapour/gas - gaseous			
1.	Melting or fusion, is a process that results in the transition of a substance from a to a			
2.	Freezing the process through which a substance changes from a to a			
	Condensation is the conversion of a or to a  Evaporation is the process in which a substance in a state changes into a state.			
solid - liquid – vapour/gas				
	Deposition, or desublimation, or reverse sublimation, is the process of solidification of a by cooling without passing through a phase.			
6.	Sublimation is a proces through which a substance changes from a to a without passing through a phase.			
В.	3. Fill in the missing proces that can occur among the three states.			
	SOLID GAS/ VAPOUR			

## C. Put in the following phrases:

 $Loosely~(x2)-~tightly-container-volume~(x3)-shape~(x2)-\\definitive-molecules~(x3)$ 

SOLID	LIQUID	GAS
A solid has a	A liquid has a	A gas takes the 10)
1) 2)	definitive	and
and its 3) Its	6) but it	10) of its
4) are5)	takes the 7) of	container. Its
packed.	its container. It has 8)	11) are very
	packed 9)	12)
		packed.